## MOLE CONVERSIONS

1) How many grams of Ar are in 2.600 moles?
2) If 1070 L of $\mathrm{CO}_{2}$ is at STP conditions, how many moles is this?
3) How many moles are in $5.7 \times 10^{24}$ atoms of calcium?
4) Convert 54.0 g of iron to moles of Fe.
5) How many atoms of zinc are equivalent to 0.403 moles of Zn ?
6) Convert 0.00333 moles of $\mathrm{O}_{2}$ gas at STP to liters.
7) 103.9 g Ar
8) $47.8 \mathrm{~mol} \mathrm{CO}_{2}$
9) 9.5 mol Ca
10) 0.967 mol Fe
11) $2.43 \times 10^{23}$ atoms Zn
12) $0.0746 \mathrm{~L} \mathrm{O}_{2}$
